

OVERVIEW

- Stable isotope labeled Internal Standards (IS) are key for accurate LC-MS metabolomics quantitation [1]. When analyzing a small number of compounds in a targeted protocol, it may be more accurate to quantify compounds individually, but when the extracts are biochemically complex, it becomes neither efficient nor economical.
- We have developed a simple workflow to efficiently quantify hundreds of metabolites in a biological extract. The workflow is suitable for both isotopic and natural abundance (NA) mixtures and sufficiently accurate for most purposes.
- A complex NA mixture is made, in which the concentration of every compound is known to establish the concentration of the same compounds in an IS mixture. The now quantified IS mixture is subsequently used to quantify unknown samples.
- In this study we examine the linearity and range of this technique.

METHODS

- In this method we used a single plate of 84 authentic from a 600 compound library (unlabeled, IROA-LSMLS) [2] that was derived from an accurately prepared mother liquors and formatted in seven 96-well plates, each well containing 1 mg of compound dissolved in 200 μ L H₂O and 50 μ L IPA (Figure 2).
- We pooled 30 μ L per well to create a solution that is serially diluted, aliquoted and dried. A fixed concentration of isotopically-labeled IS was added to all dried aliquots. Samples were analyzed using an Orbitrap Velos Pro mass spectrometer interfaced with Accela 1250 UHPLC system (Thermo Scientific). Metabolites were separated on an Ace Excel2 C18-PFP column. Apical ratios were used to calculate equivalent measurements for each compound.

WORKFLOW B (FIGURE 2)

- Each compound in the library had been previously analyzed to determine its RT. These identification parameters were loaded into a database to establish compound identity for subsequent analysis.
- The pooled serially diluted solutions containing IS were analyzed using our standard LC methodology (Figure 1). Using the database library peaks were identified and examined for their NA and IS isotopolog clusters using ClusterFinder™ software. An R program was written to perform the subsequent analysis.
- The apical ratio areas (12C/13C) were calculated and used to establish their point of equality. Since the concentrations of the 12C side were known (according to its aliquot size) therefore, the point at which the ratio was equal to 1 was used to the quantify the compounds on the IS (13C) side.
- Due to differences in the isotopic balance the ratio had to use all the isotopolog peaks for each of the NA and IS clusters in order to be accurate.
- The analysis was performed in both +ve and -ve ionization mode. This served as a secondary validation for some compounds.

Conflict of interest: Chris Beecher and Felice de Jong are co-founders of IROA Technologies

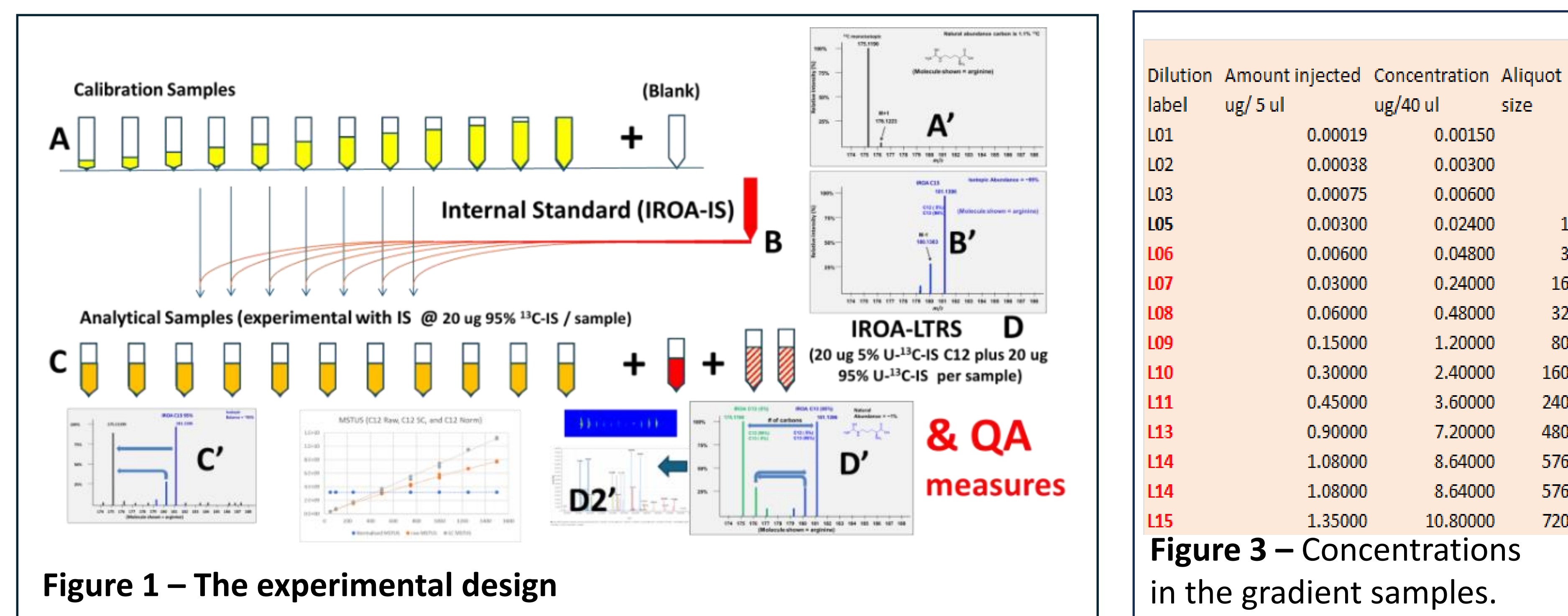


Figure 1 – The experimental design

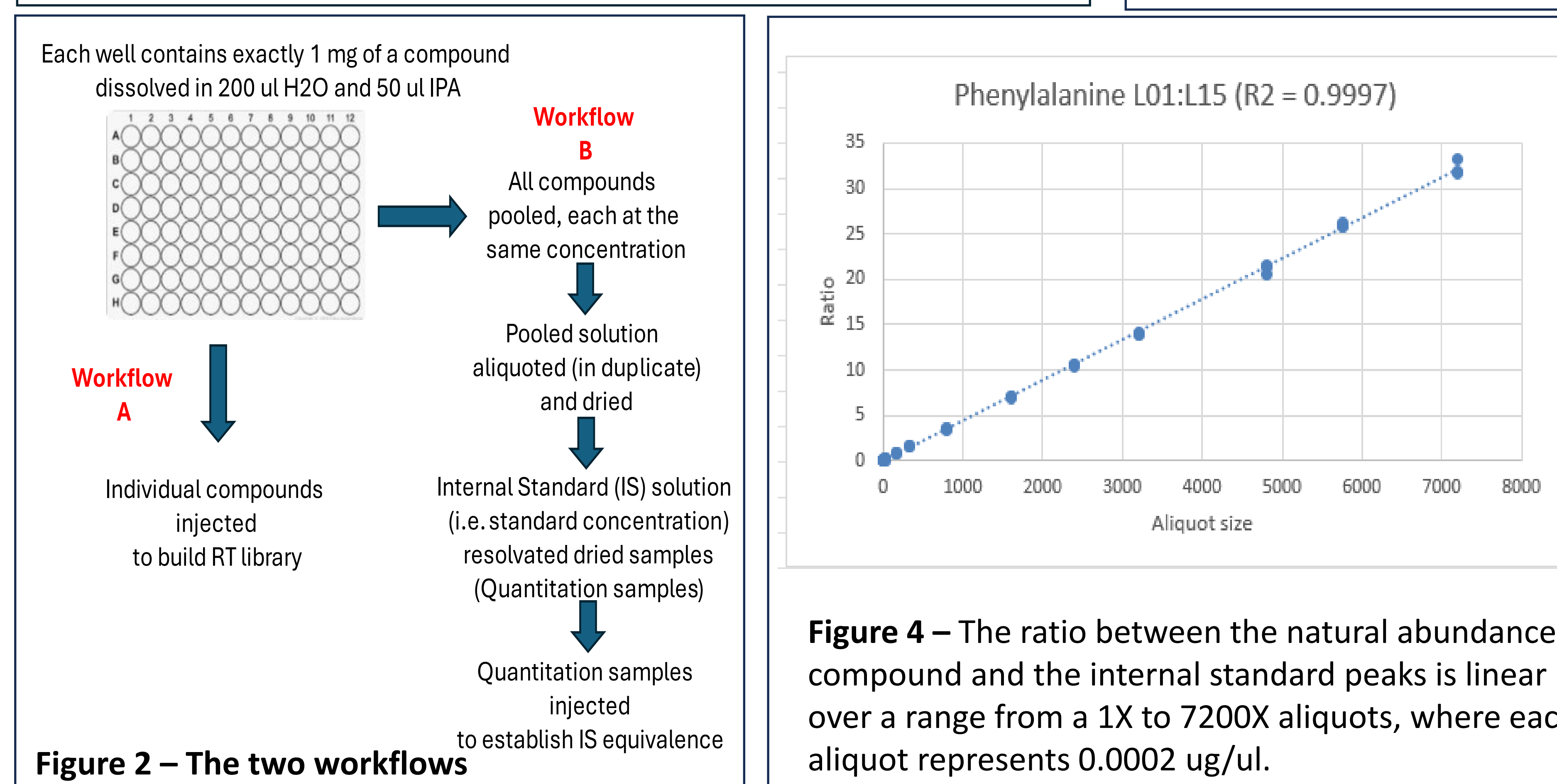


Figure 2 – The two workflows

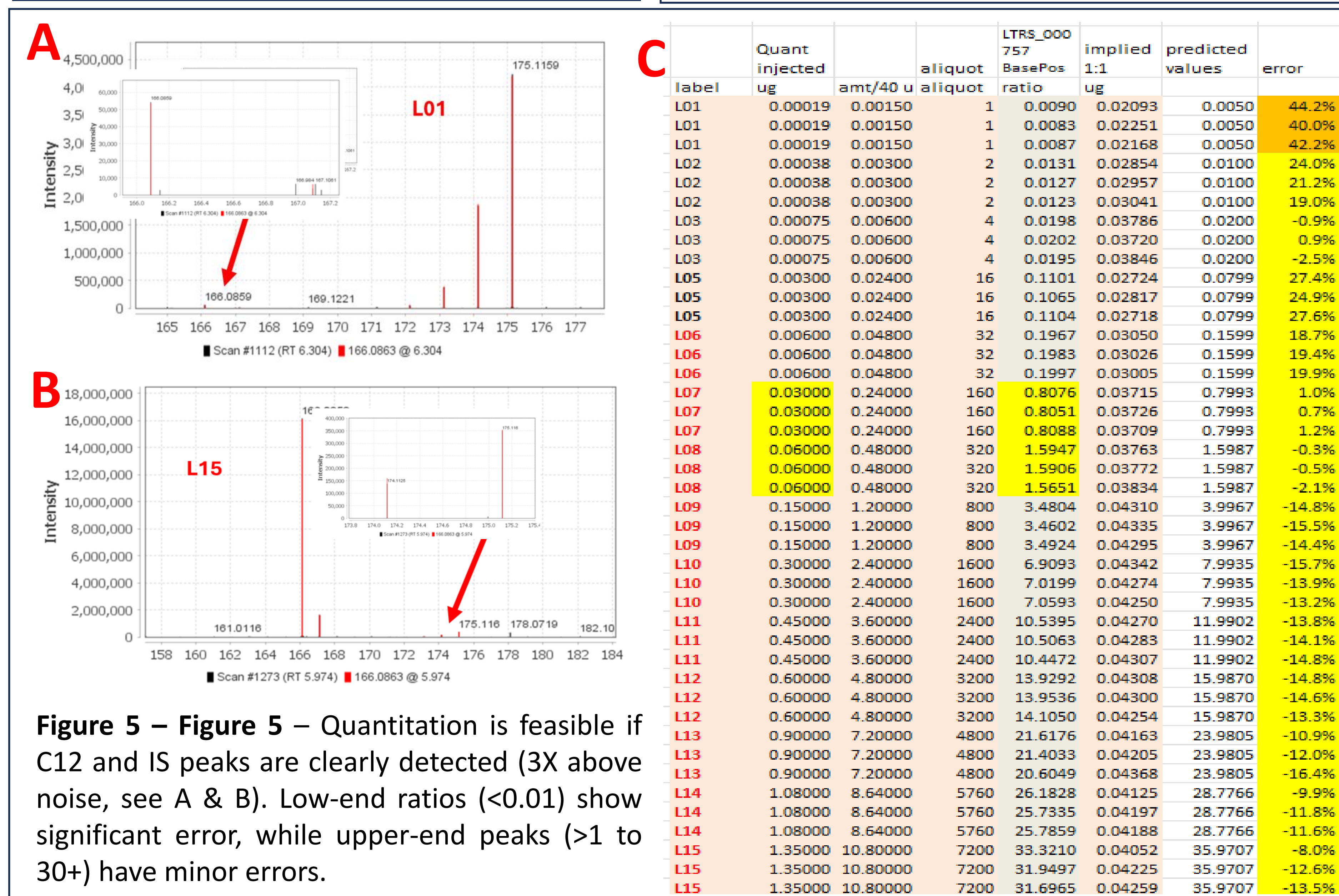


Figure 5 – Figure 5 – Quantitation is feasible if C12 and IS peaks are clearly detected (3X above noise, see A & B). Low-end ratios (<0.01) show significant error, while upper-end peaks (>1 to 30+) have minor errors.

Acknowledgements: IROA-labeled precursors for the development of IS and LTRS were produced for IROA Technologies (under agreement) by CIL.

RESULTS

This experiment developed an efficient method to quantify multiple compounds in a single mixture.

- The samples in the experimental design (see Figure 1) were analyzed according to two different workflows (see Figure 2). Both are based on a single 96 well plate which contained 84 compounds at the same concentration.
- Workflow A developed to generate a library of the exact chromatographic responses we expected to see for each compound analyzed in workflow B.
- Workflow B used a pooled mixture of all 84 compounds to create a gradient of concentrated samples (“gradient injections” in triplicate). These gradient samples were dried under a gentle nitrogen stream, and 30 μ L of a biological internal standard extract was added to resolve the samples. Thus, each gradient sample contained a known concentration of each compound and a fixed amount of internal standard.
- These resolved gradient samples were injected (5 μ L) following the library injections to keep solvent and column behavior consistent.
- The expected concentration of each compound in the “gradient” samples is shown in Figure 3. We used a per injection concentration range of 0.00019 μ g to 1.3500 μ g (Figure 3), a 7200-fold increasing gradient and a fixed concentration of IS. as shown in Figure 4.

- For each compound in each sample the apical ratio of the natural abundance (summed) peaks to the ¹³C isotopic (summed) peaks was calculated (Figure 5C). The concentration/the ratio for the samples most closely approximating 1:1 provided a measure of the amount of each compound in the internal standard. In Figure 5 the 1:1 ratio was determined to be 0.0375 μ g per injection. (highlighted in yellow)
- The error in the ratio was then back calculated for each injection (Figure 5, white and yellow columns). Within three orders of magnitude down from the 1:1 ratio the error was approximately 20% or better, and increasing concentrations from 1:1 were generally showed less than 15% error.

CONCLUSIONS

The resultant quantification of the IS makes it possible to measure hundreds of metabolites more accurately than simple relative peak comparison. This will not be as accurate as absolute quantitation of each individual compound, but it is a step forward and within the error of any measurement in a semi-targeted metabolomics analysis.

REFERENCES

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